

REB Research & Consulting

Hydrogen Separations and Membrane Reactors

12851 Capital St.
Oak Park, MI 48237
Phone: (248) 545-0155

<http://www.rebresearch.com>

25451 Gardner
Oak Park, MI 48237
Fax: (248) 545-5430

REB Polymeric Membranes

November 13, 2022

These membranes are polydimethyl siloxane; silicone, 55μ thick, 0.0055cm. The permeability of this material is understood to be the following, as measured in Barrer ($10^{-10}\text{cc}\cdot\text{cm}/\text{cm}^2\cdot\text{s}\cdot\text{cm}\cdot\text{Hg}$).

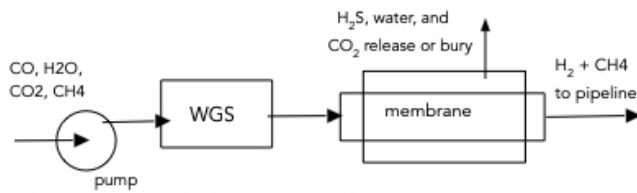
To get good separation, pick the right membrane size and use a good pressure difference between inside and out. To get a purer through-membrane gas, use a relatively small membrane area or more flow; to get a purer retentate gas, use a relatively larger membrane and less flow. For the 1000 cm^2 membranes, typical flows should be 2-5 slpm. For this module, and a partial pressure drop of 1 atm (15 psi, 76 cmHg) across the membrane, expect a through-flow of 0.83 slpm for a gas with 1000 Barrer permeability. If you need help choosing a membrane size, contact us. Some amount of consulting comes free with the membrane.

Max pressure = 60 psig for the material within the membrane tubes. The shell can withstand only about 40 psig. The maximum temperature is about 60°C . This is a polycarbonate shell, some solvents dissolve polycarbonate.

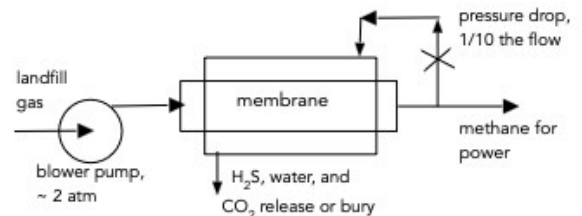
Gas	Formula	Permeability @25°C x $10^{-10}\text{cc}\cdot\text{cm}/\text{cm}^2\cdot\text{s}\cdot\text{cm}\cdot\text{Hg}$
Nitrogen	N ₂	280
Carbon monoxide	CO	340
Oxygen	O ₂	600
Nitric oxide	NO	600
Argon	Ar	600
Hydrogen	H ₂	650
Helium	He	350
Methane	CH ₄	950
Ethylene	C ₂ H ₄	1350
Ethane	C ₂ H ₆	2500
Carbon dioxide	CO ₂	3250
Propane	C ₃ H ₈	4100
Nitrous oxide	N ₂ O	4350
Acetone	C ₃ H ₆ O	5860
Ammonia	NH ₃	5900

Gas	Formula	Permeability @25°C x 10 ⁻¹⁰ cc•cm/cm ² •s•cm-Hg
Nitrogen dioxide	NO ₂	7500
Octane	n-C ₈ H ₁₈	8600
Butane	n-C ₄ H ₁₀	9000
Toluene	C ₇ H ₈	9130
Hexane	n-C ₆ H ₁₄	9400
Hydrogen sulfide	H ₂ S	10000
Benzene	C ₆ H ₆	10800
Methanol	CH ₃ OH	13900
Sulfur dioxide	SO ₂	15000
Pentane	n-C ₅ H ₁₂	20000
Water	H ₂ O	36000
Methyl Siloxane L2	..OSi(CH ₂) ₂ O..	43700
Methyl Siloxane D5	..OSi(CH ₂) ₂ O..	64300
Carbon disulfide	CS ₂	90000

The process diagrams below show how these membranes might be used to help upgrade syngas or landfill gas, or to increase the fuel efficiency of a solid oxide fuel cell.



REB suggestion: Basic process for H₂ production, if we remove less CO, this can be used to make methanol.



Basic membrane distillation used to upgrade landfill gas.